organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

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Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.003 Å R factor = 0.044 wR factor = 0.116 Data-to-parameter ratio = 15.4

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

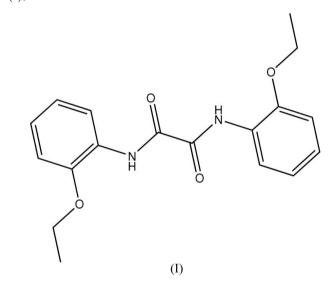
N,N'-Bis(2-ethoxyphenyl)oxamide

The title compound, $C_{18}H_{20}N_2O_4$, has an inversion center at the mid-point of the C-C bond of the oxamide unit. Intramolecular $N-H\cdots O$ and $C-H\cdots O$ hydrogen bonds make the molecule essentially planar.

Received 17 May 2006 Accepted 25 May 2006

Comment

We have recently reported the synthesis and crystal structure of N,N'-diphenyloxamide, (II) (Wen *et al.*, 2006). In our ongoing studies of oxamide derivatives, the title compound, (I), was obtained.



The compound (I) has an inversion center at the mid-point of the C-C bond of the oxamide unit (Fig. 1). The bond lengths and angles are within normal ranges (Allen *et al.*, 1987) and are comparable to those in (II). The molecule of (I) is essentially planar, with the two ethoxy groups twisted slightly to opposite sides of the molecular plane [the dihedral angle between the main molecular plane and the OCC chain is $4.15 (2)^{\circ}$]. Intramolecular hydrogen bonds (Table 1) form fiveand six-membered rings, contributing to the planarity of the molecule.

Experimental

To a solution of 2-ethoxyaniline (27.4 g, 0.2 mol) in benzene (90 ml) was added dropwise a solution of oxalyl chloride (6.4 g, 0.05 mol) in benzene (30 ml), and the mixture was stirred at 343 K for 8 h. After cooling to room temperature, water (50 ml) was added to the reaction mixture and the organic phase was washed three times with water to obtain (I) as a solid (yield 62%). Crystals of (I) suitable for X-ray analysis were obtained by slow evaporation of an N,N-dimethyl-formamide solution over a period of 9 h.

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Crystal data

 $C_{18}H_{20}N_2O_4$ $M_r = 328.36$ Monoclinic, $P2_1/c$ a = 8.7343 (10) Å b = 14.5983 (16) Å c = 7.1682 (8) Å $\beta = 110.157 (2)^{\circ}$ $V = 858.01 (17) \text{ Å}^3$

Data collection

Siemens SMART 1000 CCD area
detector diffractometer
ω scans
Absorption correction: multi-scan
(SADABS; Sheldrick, 1996)
$T_{\rm min} = 0.972, T_{\rm max} = 0.989$

Refinement

$\begin{array}{ll} \mbox{Refinement on } F^2 & w = 1/[\sigma^2(F_o^2) + (0.0588P)^2 \\ & w = 1/[\sigma^2(F_o^2) + (0.0588P)^2 \\ & + 0.0318P] \\ & w = P = (F_o^2 + 2F_c^2)/3 \\ S = 1.05 & (\Delta/\sigma)_{max} < 0.001 \\ 1675 \ reflections & \Delta\rho_{max} = 0.13 \ e \ \text{\AA}^{-3} \\ 109 \ parameters & \Delta\rho_{min} = -0.14 \ e \ \text{\AA}^{-3} \\ \mbox{H-atom parameters constrained} \end{array}$

Table	1
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Hydrogen-bond geometry (Å, $^{\circ}$).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$N1-H1A\cdots O1$ $N1-H1A\cdots O2^{i}$ $C7-H7A\cdots O2$	0.86	2.15	2.576 (2)	110
	0.86	2.23	2.668 (2)	112
	0.93	2.39	2.976 (2)	121

Z = 2

 $D_x = 1.271 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation

Column, colorless

 $0.32\,\times\,0.16\,\times\,0.12$ mm

4858 measured reflections 1675 independent reflections 1208 reflections with $I > 2\sigma(I)$

 $\mu = 0.09 \text{ mm}^{-1}$ T = 293 (2) K

 $R_{\rm int} = 0.023$

 $\theta_{\rm max} = 26.0^\circ$

Symmetry code: (i) -x + 1, -y + 1, -z + 1.

All H atoms were located in difference Fourier maps and constrained to ride on their parent atoms, with C-H = 0.93-0.97 Å, N-H = 0.86 Å and $U_{iso}(H) = 1.2U_{eq}(C,N)$ and $1.5U_{eq}(methyl C)$.

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINT* (Siemens, 1996); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 1997); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*, *PARST* (Nardelli, 1995) and *PLATON* (Spek, 2003).

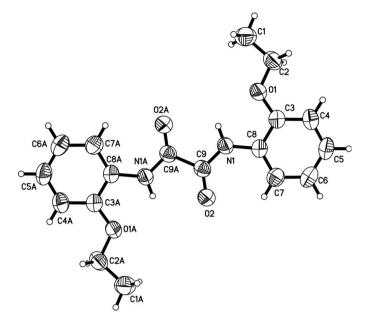


Figure 1

The structure of (I), showing 50% probability displacement ellipsoids and the atom-numbering scheme. The suffix A corresponds to symmetry code (1 - x, 1 - y, 1 - z).

This project was supported by the Special Project of Qingdao for Leadership of Science and Technology (No. 05–2-JC-80) and the Outstanding Adult-Young Scientific Research Encouraging Foundation of Shandong Province (No. 2005BS04007).

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